

EMA 3011 – Fundamental Principles of Materials – Spring 2017
Section 9765

1. **Course Description:** The fundamental principles of structure, reactivity and energies describing materials systems will be covered, directly relating individual principles to specific materials properties or functions. (3 credit hours)

2. **Course Objectives:** In this course the student is introduced to the way in which the fundamentals of organic materials and the physical laws of quantum mechanics influence materials properties and reactions.

Specific Objectives:

- Become familiar with the fundamentals of organic chemistry and be able to apply them to the chemical and physical properties and processing of polymer materials
- Learn the laws of quantum mechanics and understand the way in which they influence materials properties

3. **Prerequisites:** CHM 2046 or CHM 2096

4. **Contribution of course to meeting the professional component:** This course provides 3 credits towards engineering sciences.

5. Instructor: **Dr. Jennifer Andrew**

- a. Office location: **162 Rhines Hall**
- b. Telephone: **352.846.3345**
- c. E-mail address: **jandrew@mse.ufl.edu**
- d. Office hours: **R 10:30 am – 12:00 pm**
- e. Website: **<https://elearning.ufl.edu/>**

6. Teaching Assistant: **TBD**

- a. Office hours location: **TBD**
- b. E-mail address: **TBD**
- c. Office hours: **TBD**

7. Meeting Times and Location: **T Period 2, 8:30 am – 9:20 am Florida Gym 270**

R Period 2 & 3, 8:30 am – 9:20 am & 9:35 am – 10:25 am

8. Textbooks Required

- 1a. Title: ***Introduction to Organic Chemistry***
- 1b. Author: **William Brown and Thomas Poon**
- 1c. Edition: **Fifth Edition (John Wiley & Sons, Inc.)**
- 1d. ISBN: **978-0-470-12923-4**

- 2a. Title: ***Elements of Physical Chemistry***
- 2b. Author: **Peter Atkins and Julio de Paula**
- 2c. Edition: **Fifth Edition (Freeman)**
- 2d. ISBN: **1-4292-1813-4**

9. On the Web: This course will use Canvas extensively as a communication and archival tool. The students can access all relevant course information (course notes, homework and exam solutions, announcements, grades, etc.) via the Canvas entry link: <https://elearning.ufl.edu/>. Pertinent course information may also be announced via UFL e-mail address in addition to over Canvas.

10. Conduct, Attendance and Expectations: Proper behavior in class is always important and leads to a relaxed and productive educational environment. Thus, eating, drinking, texting, reading of newspapers, working on homework for this or other courses, or other activities that are not part of the class are not allowed. Students who do not comply with these requirements or who behave disorderly or disrespectfully may be asked to leave the classroom. Leaving your cell phone on, leaving early or arriving late can be VERY distracting. All electronic devices (cell-phones, etc.) should be turned off or in silent mode. If your cell phone rings during class it will be confiscated for the remainder of the class period. While not directly enforced, attendance is strongly recommended for this course. While attendance does not make up a specific component of the course grade, it will be reflected in homework and exam grades.

11. **Grading and Grading Scale:** Your final grade will be allocated based on the following distribution:

Homework: 25%
 Three In-Class exams: 25% each

Percentage	≥ 92	≥ 88	≥ 84	≥ 80	≥ 76	≥ 72	≥ 68	≥ 65	≥ 62	≥ 59	≥ 56	< 56
Letter Grade	A	A-	B+	B	B-	C+	C	C-	D+	D	D-	E

The instructor reserves the right to adjust the grade distributions, however grades **will not** be adjusted for individuals.

Exams. You will be given 3 exams throughout the semester, the exam content may change but the dates will not. **There will be no final exam.** Each exam is weighted equally and will be worth 25% of your final grade. Requests for re-grading must be made in writing and within one week after an assignment has been returned. **Make-up exams** will be provided only with the *prior approval of the instructor or excused absence*. In general, acceptable reasons for excused absence include illness, serious family emergencies, special curricular requirements, military obligation, court-imposed legal obligations, religious holidays and participation in official university activities such as music performances, athletic competition or debate.

Calculators. Only scientific and non-graphing calculators will be permitted during exams. The permitted calculators are NCEES (National Council of Examiners for Engineering and Surveying) approved, and include the following:

- Hewlett Packard- HP 33s and HP 35s
- Casio – fx-115 ES, fx-115 MS, fx-115 MS Plus, fx-115 MS SR
- Texas Instruments – TI-30Xa, TI-30Xa SOLAR, TI-30Xa SE, TI-30XS Multiview, TI-30X IIB, TI-30X IIS, TI-36X II, and TI-36X SOLAR, TI 36X Pro
- **If you are unsure about your calculator, it is your responsibility to check with the instructor for approval. Calculators will not be provided if you forget or bring the wrong calculator.**

Homework. Homework problems from the back of each chapter will be assigned, together with due dates, through e-Learning. These homework questions are *essential* to your study and some exam questions will be adapted from them. Homework will be posted through the Canvas web site. Assignments are due at the **beginning of class (8:30 am)** on the due date, unless otherwise stated. **Late homework assignments will not be accepted and will receive a grade of zero.** The lowest homework score will be dropped.

12. **Course Outline:** Below is the tentative schedule of topics, activities, reading assignments, and exams.

Week of	Topic	Reading Assignments
Jan. 2	Introduction, Covalent Bonding	B&P Ch. 1
Jan. 9	Alkanes, Cycloalkanes	B&P 3.1-3.5
Jan. 16	Alkanes (cont.), Chirality	B&P 3.6, 3.7, 3.9, 6.1-6.5
Jan. 23	Chirality (cont.), Alkynes & Alkynes	B&P 4.1-4.3
EXAM 1	Thurs. Feb. 2	
Jan. 30	Exam Review	
Feb. 6	Reactions of Alkenes	B&P 5.3-5.6
Feb. 13	Reactions of Alkenes (cont.), Alkyne Reactions	B&P 5.7-5.8
Feb. 20	Haloalkanes	B&P 7.1-7.9
Feb. 27	Alcohols & Ethers, Benzene & its Derivatives	B&P 8.1-8.6, 9.1-9.8
Mar. 6	<i>Spring Break</i>	
EXAM 2	Thurs. March 16	
Mar. 13	Exam Review	
Mar. 20	Introduction to Quantum Mechanics	A & de P Ch. 12.1-12.6
Mar. 27	Applications of Quantum Mechanics, Quantum Dots	A & de P Ch. 12.7, Supp.
Apr. 3	Rotational Motion, Hydrogenic Atom	A & de P Ch. 12.8, Ch. 13
Apr. 10	Hydrogenic Atom, Exam	A & de P Ch. 13
EXAM 3	Thurs. April 13	
Apr. 12	Exam Review	
Apr. 17	Quantum Numbers	A & de P Ch. 13

14. **Honesty Policy** – All students admitted to the University of Florida have signed a statement of academic honesty committing them to be honest in all academic work and understanding that failure to comply with this commitment will result in disciplinary action. This statement is a reminder to uphold your obligation as a UF student and to be honest in all work submitted and exams taken in this course and all others.

15. **Accommodation for Students with Disabilities** – Students Requesting classroom accommodation must first register with the Dean of Students Office. That office will provide the student with documentation that he/she must provide to the course instructor when requesting accommodation.

16. **UF Counseling Services** – Resources are available on-campus for students having personal problems or lacking clear career and academic goals. The resources include:

- University Counseling Center, 301 Peabody Hall, 392-1575, Personal and Career Counseling.
- SHCC mental Health, Student Health Care Center, 392-1171, Personal and Counseling Center for Sexual
- Assault/Abuse Recovery and Education (CARE), Student Health Care Center, 392-1161, sexual assault counseling.
- Career Resource Center, Reitz Union, 392-1601, career development assistance and counseling.

17. **Software Use** – All faculty, staff and student of the University are required and expected to obey the laws and legal agreements governing software use. Failure to do so can lead to monetary damages and/or criminal penalties for the individual violator. Because such violations are also against University policies

and rules, disciplinary action will be taken as appropriate. We, the members of the University of Florida community, pledge to uphold ourselves and our peers to the highest standards of honesty and integrity.